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Example 6. Calculate The Enthalpy Change Of Reaction For The Reaction Where 25.0cm³ Of 0.20 Mol Dm⁻³ Copper Sulfate Was Reacted With 0.01mol (e Xcess Of Zinc). The Temperature Increased 7.0oC . Step 1: Calculate The Energy Change For The Amount Of Reactants In The Calorimeter. $Q = M \times C_p \times T$ $Q = 25 \times 4.18 \times 7$ $Q = 731.5 \text{ J}$ Step 2 : Calculate The Number Of Moles Of The Reactant Not In Excess ... Mar 8th, 2021

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